

• Hydrogen Atom / Coulomb Potential

- Basics:

- The potential is $V(r) = -\frac{e^2}{4\pi\epsilon_0 r}$ and is central

Therefore we know angular wave function is spherical harmonic $Y_l^m(\theta, \phi)$

- Radial equation is (recall $u = rR(r)$)

$$-\frac{\hbar^2}{2m} \frac{d^2u}{dr^2} + \left[-\frac{e^2}{4\pi\epsilon_0 r} + \frac{\hbar^2}{2m} \frac{l(l+1)}{r^2} \right] u = Eu$$

(As argued on HW, m =electron mass) should be replaced by μ =reduced mass

- The energy eigenvalues will be (bound states)

$$E_n = -\frac{\mu}{2\hbar^2} \left(\frac{e^2}{4\pi\epsilon_0} \right)^2 \frac{1}{n^2} \quad \text{radial scale set by}$$

$$\text{Bohr radius } a = \frac{4\pi\epsilon_0 \hbar^2}{me^2}$$

+ States are therefore

$|N, l, m; s=1/2, m_s\rangle$ counting electron spin (add $|s=1/2, m_s\rangle$)

or $|N, j, m, l, s=1/2\rangle$

\rightarrow total angular momentum
for proton in nucleus

+ Energy is independent of m, m_s due to invariance under rotations

But it is also independent of l (or j). Special to Coulomb potential.

- Radial Equation (bound states only $E < 0$)

- Dimensionless form

+ Define $\gamma \hbar = \sqrt{-2\mu E}/\hbar$ (as before) Then $\rho = \gamma \hbar r$ is dimensionless

+ Radial equation is now dimensionless

$$\frac{d^2u}{d\rho^2} = \left[1 - \left(\frac{mc^2}{4\pi\epsilon_0 \hbar^2 k} \right) \frac{1}{\rho} + \frac{l(l+1)}{\rho^2} \right] u$$

You can simplify a bit by calling $P_0 = mc^2/2\pi\epsilon_0 \hbar^2 k = 1/ka$

- Asymptotics

+ As $\rho \rightarrow 0$, we recall $u \propto \rho^{l+1}$ for normalizability (centrifugal term dominates at $l \geq 1$).

- + As $\rho \rightarrow \infty$, $d^2u/d\rho^2 = u \Rightarrow u \propto e^{-\rho}$ for normalizable behavior
up to ρ_0 , up to polynomial corrections.
- Series solution for $v(\rho)$
 - + Guess at form $u(\rho) = \rho^{l+1} e^{-\rho} v(\rho)$ (even for $l=0$)
 - + Radial equation becomes
$$\rho \frac{d^2v}{d\rho^2} + 2(l+1-\epsilon) \frac{dv}{d\rho} + (\rho_0 - 2(l+1))v = 0$$

(Please check the book for yourself)
 - + Set $v(\rho) = \sum_{j=0}^{\infty} c_j \rho^j$ (cannot have - powers b/c of $\rho \rightarrow 0$ asymptotics)
 - After some algebra (see text again), get recursion relation
$$c_{j+1} = \frac{2(j+l+1) - \rho_0}{(j+1)(j+2l+2)} c_j$$
 - + The series must terminate (or it changes $e^{-\rho}$ form of $\rho \rightarrow \infty$ asymptotic)
This means some j_0 is largest w/nonzero coeff.
 $\Rightarrow 2j_0 + 2l + 2 = \rho_0 = 2n$ where $n = 1, 2, 3, \dots, n \geq l+1$
- Alternate solutions b/c we don't want to use recursion all day.
 - + The differential eqn $xv'' + (v+1-x)v' + \lambda v = 0$ is associated Laguerre eqn.
By comparison, this is our radial eqn if $\rho \propto x$
 $\rho = x/2, v = 2l+1, \lambda = n-l-1$
The normalizable solution is associated Laguerre polynomial $L_{n-l-1}^{2l+1}(2\rho)$
See text for some properties: In particular, $\lambda \geq 0$, integer, far normalizable.
 - + May also use an algebraic procedure similar to harmonic oscillator but more complicated + less illuminating.
See Ohanian.

- Wavefunctions + Spectra

- Bohr radius + Bohr energy
- + Normalizability requires $\rho_0 = 2n \Rightarrow K = 1/na$, $a = \frac{4\pi e^2 k^2}{mc^2}$
- + The definition of $2K$ gives the Bohr formula

$$E = -\frac{\hbar^2}{2ma^2} \frac{1}{n^2} = -\frac{m}{2\hbar^2} \left(\frac{e^2}{4\pi\epsilon_0}\right)^2 \frac{1}{n^2}$$

- + Transitions from n_i to n_f produce light of wavelength λ (near visible) but $\frac{hc}{\lambda} = E_{n_f} - E_{n_i}$ or $\frac{1}{\lambda} = R \left(\frac{1}{n_f^2} - \frac{1}{n_i^2} \right)$, R = Rydberg constant
- The wave function assembles all the parts
 - + $\rho = Kr = r/na$, so $R = 4/r \propto (r/na)^l e^{-r/na} L_{n-l-1}^{2l+1}(2r/na)$
 - + The angular parts are spherical harmonics (and spin stuff)
 - + $\psi_{nlmm_s} = \sqrt{\left(\frac{2}{na}\right)^3 \frac{(n-l-1)!}{2^n (n+l)!}} e^{-r/na} \left(\frac{r}{na}\right)^l L_{n-l-1}^{2l+1}\left(\frac{2r}{na}\right) Y_l^m(\theta, \phi) |m_s\rangle$
 - + Energy depends only on principal quantum number n , but wave functions depend on all of them.